

CHEM 1100/1102: OUTLINE OF GENERAL CHEMISTRY

Fall 2023

LECTURE A: CHEM 1100 (90035) & CHEM 1102 (90471), M,W,F 8:30AM-9:20AM, Innovation E102

LECTURE B: CHEM 1100 (90626) & CHEM 1102 (90472), T,Th 8:30AM-9:45AM, Votey 105

GENERAL INFORMATION: (see also the CHEM 1100 BrightSpace page)

Instructor: Steve Flemer **email:** sflemer@uvm.edu

Office: 331 Innovation

Office Hours: Mon, Wed, Fri: 9:30-10:30AM In-person: 331 Innovation
Tue, Thurs: 10:00-11:00AM In-person 331 Innovation

If the following office hour times do not work for your schedule, please email me to try to set up a virtual office hours session on Teams.

Lecture: The lecture will primarily be used to cover new material. Included in this syllabus is a tentative schedule covering the class material and the general flow of how the course is laid out and assessed.

Exams: Three 2-hour exams are given on Thursday nights from 6:00-8:00 PM.

	Lecture A (M,W,F section)	Lecture B (T,Th section)
Exam 1	Thurs, Sept. 21; Billings Lecture Hall	Thurs, Sept. 21; 301 Williams
Exam 2	Thurs, Oct. 19; Billings Lecture Hall	Thurs, Oct. 19; 301 Williams
Exam 3	Thurs, Nov. 16; Billings Lecture Hall	Thurs, Nov. 16; 301 Williams
Final Exam	Thurs, Dec. 14; 7:30-10:15AM; Innovation E102	Tues, Dec. 12; 7:30-10:15AM; 105 Votey

Absences from exams: Makeup exams will only be administered after the scheduled exam time at the instructor's discretion if a **medical or family emergency** or a **UVM-Related Conflict** (athletic commitment, class conflict, etc.) precludes taking the exam at the scheduled time. This must be cleared with me first, however. Authorized makeup exams must be carried out within a week of the scheduled exam.

Review Sessions: I will be holding Exam Review Sessions the Wednesday evenings prior to impending exams. Weekly SI sessions will also be starting shortly after the beginning of classes. Firm dates for these Review Sessions and SI sessions will be announced.

Problems: Exam questions will be modeled very closely to the type of problems you will encounter in the Practice Problems of each unit of study posted on BrightSpace. Solutions to all of these problems are included in these documents. While it is strongly suggested that you do as many problems as possible, the problems are not collected or graded.

Weekly BrightSpace Quizzes: Each week, you will be responsible for taking a short online BrightSpace quiz covering the class material from the current week. Just click on the “Weekly Quiz” link on the left-hand side of the Content & Activities area of the CHEM 1100 BrightSpace page and follow the instructions. These quizzes are open-book, but must be completed independently. Weekly quizzes will be available to take until midnight of the Sunday prior to a new week of classes. A skipped or a missed quiz is given a zero.

REQUIRED COURSE MATERIALS:

Text: There is no textbook for the course. Each unit of study has a corresponding folder in the Course Materials section in the Content & Activities area of the course’s BrightSpace site, within which are educational notes for that unit. These notes, while helpful for following along with the material, should not be thought of as comprehensive. Your own written class notes should be the basic core of your study materials.

Scientific Calculator: A standard scientific calculator is a requirement for the exams.
Note: Graphing calculators are not allowed on exams.

Lab Manual: Available for download from the class’ BrightSpace site.

Bound Laboratory Notebook: Available at the UVM Bookstore. Required for recording data.
(Note: the last two items are not required for CHEM 1102 students)

Inclusion Statement: I want everyone to be successful and fulfilled in this course. As such, I do not play favorites and treat every student with the same respect we all deserve from one another regardless of who you are, what you look like, and what your beliefs are. We are here to master general chemistry, and I will do everything in my power toward your success in the course.

Academic Integrity: Offenses against the Code of Academic Integrity (ie: Cheating) are deemed serious and insult the integrity of the entire academic community. Any suspected violations of the code are taken very seriously and will be forwarded to the Center for Student Ethics & Standards for further investigation.

LABORATORY:

(labs start 2 weeks after classes begin)

Time and Room: See your class course schedule as to your assignments.

Attendance: Students must attend the lab section they are assigned to. Official documentation of sickness or family crisis is required if a lab is missed. **If more than 2 labs are missed, this results in a failure for the course.** In order to take a lab at a time other than your assigned time one must obtain the permission of the TA and instructor.

Online Lab Safety Quiz: Prior to the lab sessions beginning, students must read through Lab Safety documentation and take a one-time online quiz before being allowed into their lab session. Just click the “Lab Safety” link on the left hand side of the Content & Activities area of the CHEM 1100 BrightSpace page and follow the instructions. Students must score an 80 or better on the quiz to be admitted to lab. If you choose, you may take the Lab Safety quiz as many times as you want in order to maximize this score, as it will also count as your first lab quiz grade.

Lab Safety considerations & Gear: OSHA approved safety glasses and disposable lab coat (both available at the UVM Bookstore) must be worn by everyone once any experimentation has started in any area of a lab room. Only shoes that cover the toes are permitted in the lab. Sandals and open-toed shoes are not permitted.

Lab Schedule:

<u>Date</u>	<u>Experiment Description</u>
11 – 13 SEPT	<u>CHECK-IN</u> & Densities of Common Substances
18 - 22 SEPT	Determination of Heat Capacity Using Calorimetry
25 - 27 SEPT	Qualitative Analysis
2 - 4 OCT	Synthesis of Ionic Compound Alum from Aluminum Metal
9 - 11 OCT	Determination of a Compound’s Empirical Formula
16 - 18 OCT	Reaction Stoichiometry & Equation Balancing
23 - 25 OCT	Determination of Limiting Reactant
30 OCT - 1 NOV	Determination of Acid Content in Pickle Juice using Titration
6 - 8 NOV	Determination of Limestone Content in Soil using the Ideal Gas Law
13 – 15 NOV	Acid-Base Equilibria and Buffers & <u>LAB CHECK-OUT</u>

COURSE GRADE FOR CHEM 1100 STUDENTS:

1. Points needed to obtain a specific grade

920 = A	870 = B+	790 = B-	680 = C	620 = D+	570 = D-
900 = A-	820 = B	760 = C+	650 = C-	590 = D	less than 570 = F

Class = Out of 800 possible points:

Exam 1:	100	} (I will drop your lowest semester exam/quiz score before calculating course grade) excluding lowest score	= 300 points possible for semester tests
Exam 2:	100		
Exam 3:	100		
Quiz Average:	100		
Final Exam:	100		+ 100 points possible for final exam
			<hr/>
			400 points
			(x 1.75) = 700 points possible for all tests/quizzes
Attendance:	100		+ 100 points possible for attendance
			<hr/>
			= 800 points possible for lecture points

The 1.75 multiplication factor is because each graded component is only worth 100 pts, and thus the maximum number of points obtainable from the tests is 400. In order to raise this to 700 possible points you must multiply the total x 1.75. Attendance (possible 100) are then added in afterwards to give a highest possible score of 800 for lecture points.

Laboratory: = 200 points possible for lab points

Example Course Grade:

	<u>Exam 1</u>	<u>Exam 2</u>	<u>Exam 3</u>	<u>Quiz av</u>	<u>Final.</u>	<u>Attendance</u>
Scores	85	45	78	77	75	80
		(dropped)				

Total test/quiz points = 315 (x 1.75) = 551.25

+ 80 attendance points: **631.25 total lecture points**

Lab points = 160 + **160 Lab points**

791.25 total points = B- for the course

COURSE GRADE FOR CHEM 1102 STUDENTS:

Since there is no lab component to your grade, you will be graded on your test and attendance scores exclusively. Your 4 highest test scores will be multiplied by 2.25 (rather than 1.75), with attendance grade added in afterwards to equal 1000 possible points.

TENTATIVE LECTURE SCHEDULE

UNIT 1 (Measurement & Problem Solving)

UNIT 2 (Matter & Energy)

UNIT 3 (Atoms & Elements)

Exam 1 (Thursday, Sept. 20; 6:00-8:00PM)

UNIT 4 (Electrons in Atoms)

UNIT 5 (Chemical Bonding)

UNIT 6 (Molecules & Compounds)

UNIT 7 (Chemical Composition)

Exam 2 (Thursday, Oct 19; 6:00-8:00PM)

UNIT 8 (Chemical Reactions)

UNIT 9 (Quantities in Chemical Reactions)

UNIT 10 (Solutions)

UNIT 11 (Gases)

UNIT 12 (Liquids, Solids, & Intermolecular Forces)

Exam 3 (Thursday, Nov 16; 6:00-8:00PM)

UNIT 13 (Chemical Equilibrium)

UNIT 14 (Acids & Bases)

Final Exam (Cumulative)